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8 OXIDATION N REACTION 1

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Addison For each reaction below, identify
the atom oxidized, the Page 5/21. Read

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Addison Wesley atom reduced, the

oxidizing agent, and the reducing agent. 1)

$Mg + 2HCl (MgCl_2 + H_2)$ 2) $2Fe +$

$3V_2O_3 (Fe_2O_3 + 6VO)$ 3) $2KMnO_4 +$

$5KNO_2 + 3H_2SO_4 (2MnSO_4 + 3H_2O +$

$5KNO_3 + K_2SO_4$ Oxidation-Reduction

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Chapter 6 Oxidation-Reduction Reactions

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